

# **Development of a Sampling and Analysis Method for the Measurement of Hexavalent Chromium in Ambient Air**

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# Background

- Hexavalent chromium [Cr(VI)] is recognized as a pulmonary carcinogen. Cr(VI) compounds have been enlisted as one of the 18 core HAPs by the USEPA.
- It is difficult to measure Cr(VI) because of its instability.
  - $\text{Cr(VI)} \leftrightarrow \text{Cr(III)}$
- There was no standard method available for the measurement of Cr(VI) in air.

# EPA Method for Cr(VI) Measurement in Air

- Developed by Eastern Research Group (ERG)
  - IC-UV method
    - NaHCO<sub>3</sub> pre-treated cellulose filter for collection
    - IC separation
    - post-column derivatization with diphenylcarbohydrazide
    - UV detection at 540 nm
  - MDL: 0.0074 ng/m<sup>3</sup>

## Limitations of the EPA Method

- **Can't monitor Cr(VI) ↔ Cr(III) inter-conversion.**
- **Stability of Cr(VI) during sampling and analysis has not been thoroughly evaluated.**

# Objectives

**Develop a reliable, sensitive sampling and analytical method to measure hexavalent chromium [Cr(VI)] in ambient air.**

- **Optimize the Ion Chromatography-Inductively Coupled Plasma Mass Spectrometry (IC/ICPMS) method for Cr(VI) analysis and lower the detection limit.**
- **Investigate the optimal extraction condition for Cr(VI) air samples.**
- **Investigate the stability of Cr(VI) during sampling, sample processing and sample storage after sampling.**
- **Quantify Cr(VI) and Cr(III) interconversion using the EPA 6800 method (Speciated isotope-dilution mass spectrometry method)**

# Flow Chart for Cr(VI) Measurement

Air Sampling  
(Sampling Artifact)



Sample storage  
(Stability)

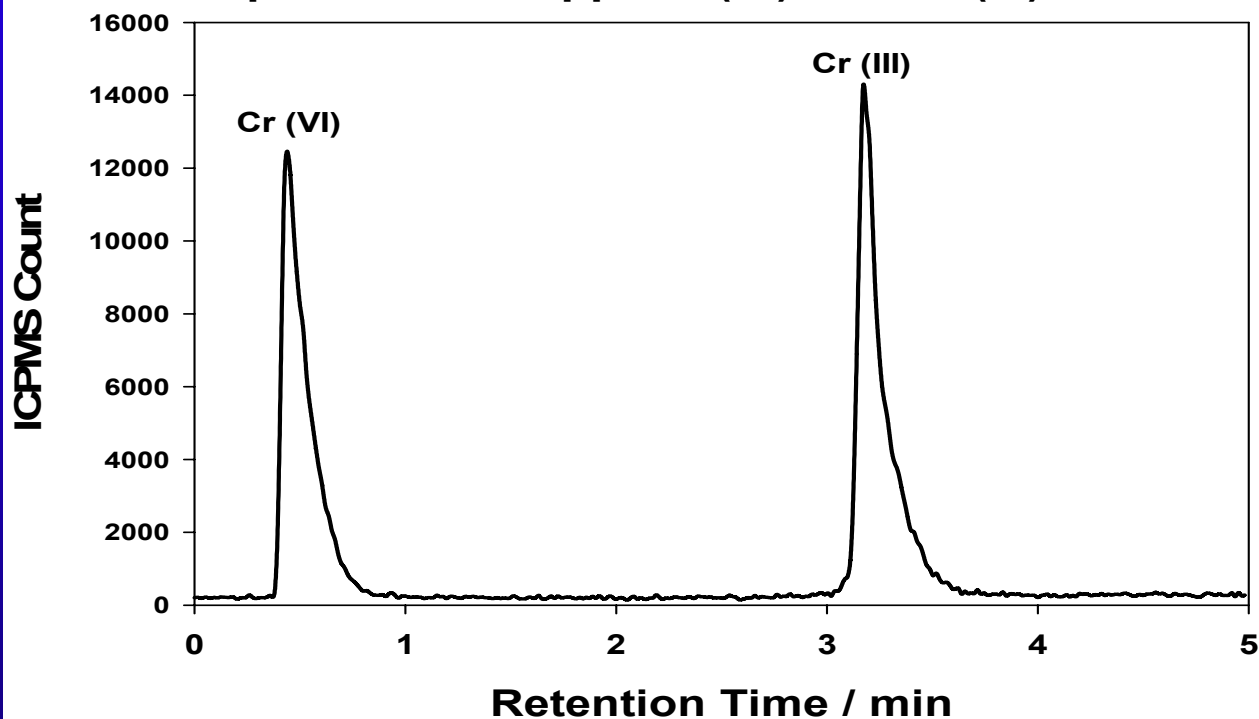


Extraction  
(Stability and Efficiency)

Ion Chromatography/Inductively Coupled Plasma  
Mass Spectrometry (IC/ICPMS)  
(Optimal separation condition for IC; Optimal  
detection condition for ICPMS)

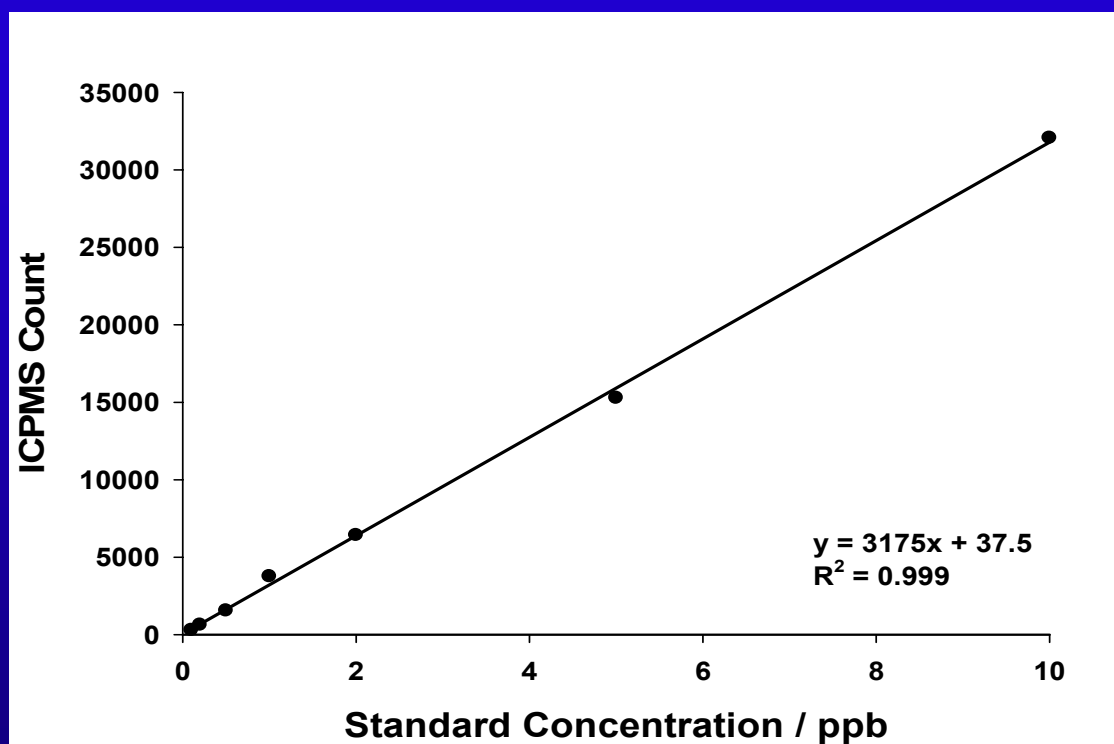


## Spectrum of 20ppb Cr(VI) and Cr (III)



- Separation Condition:
- CG5A cation exchange column
- HNO<sub>3</sub> as mobile phase with flow rate of 1.25 mL/min
- IC program:
- Injection volume: 100 μL/min  
60% 1M HNO<sub>3</sub> and 40% DI H<sub>2</sub>O

# Instrument Tuning and Detection Limit



Tuning Solution	Analytical Detection Limit	Equivalent to Air Conc.(24 m <sup>3</sup> )
10 ppb Indium	0.12 ppb	0.050 ng/m <sup>3</sup>
10 ppb <sup>52</sup> Cr	0.32 ppb	0.13 ng/m <sup>3</sup>
10 ppb <sup>53</sup> Cr	0.25 ppb	0.10 ng/m <sup>3</sup>



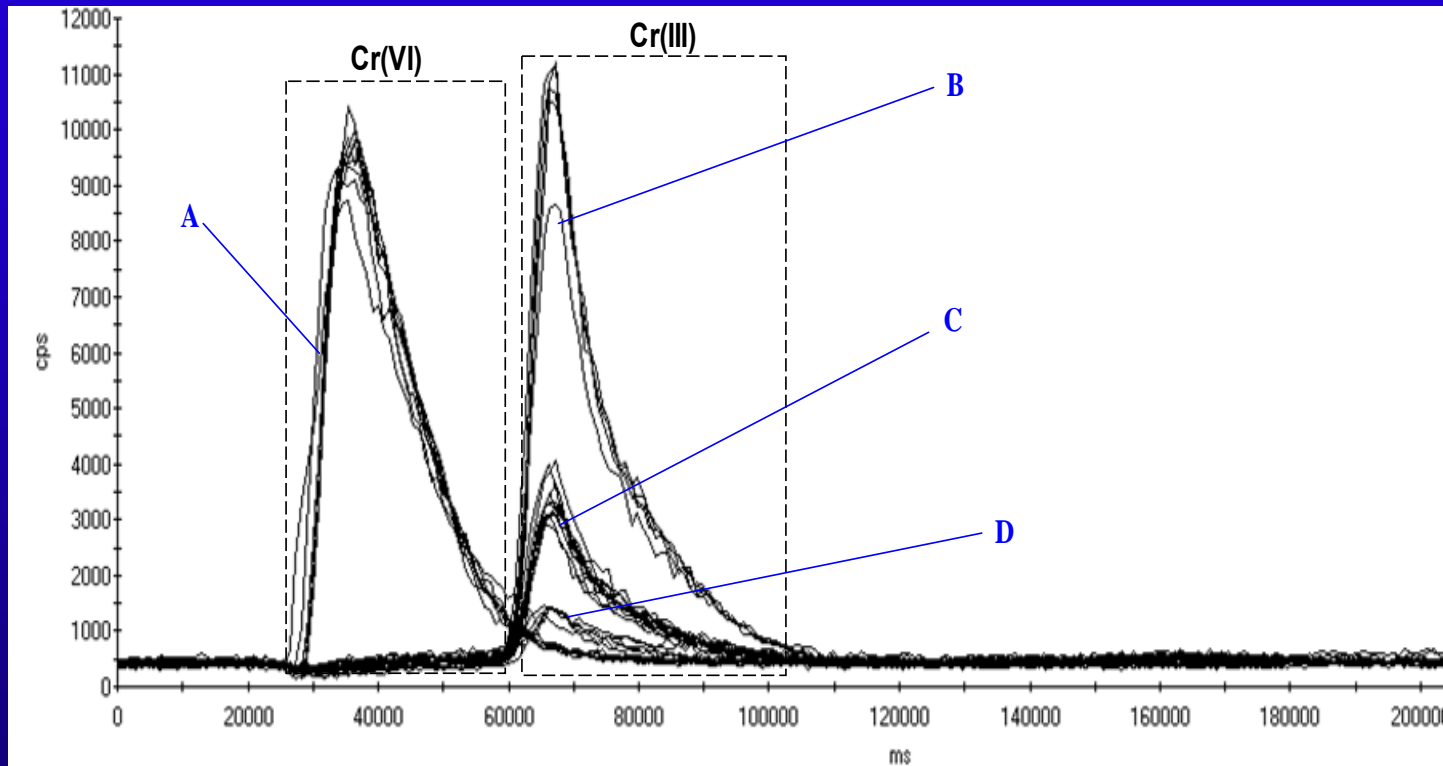
# Optimal Extraction Method

- No Interferences
- High extraction efficiency
- Cr(VI) is stable during extraction
- Minimum interconversion between Cr(VI) and Cr(III)

# Extraction Solution

- Extraction solutions tested:
  - $\text{H}_2\text{O}$
  - $\text{NaHCO}_3/\text{Na}_2\text{CO}_3$
  - $\text{NaOH}$
  - $\text{NH}_4\text{OH}$
  - $\text{K}_2\text{HPO}_4/\text{KH}_2\text{PO}_4$
  - $\text{NH}_4\text{H}_2\text{PO}_4/(\text{NH}_4)_2\text{HPO}_4$
  - $\text{HNO}_3$
- Sample tested: NIST 1648 particulate matter + Cr(VI) + Cr(III) + Teflon or  $\text{NaHCO}_3$  pretreated cellulose filter.

# Chromatogram of Cr(VI) and Cr(III)



Group A: Cr(VI) under neutral, acidic and basic conditions

Group B: Cr (III) under neutral and acidic conditions.

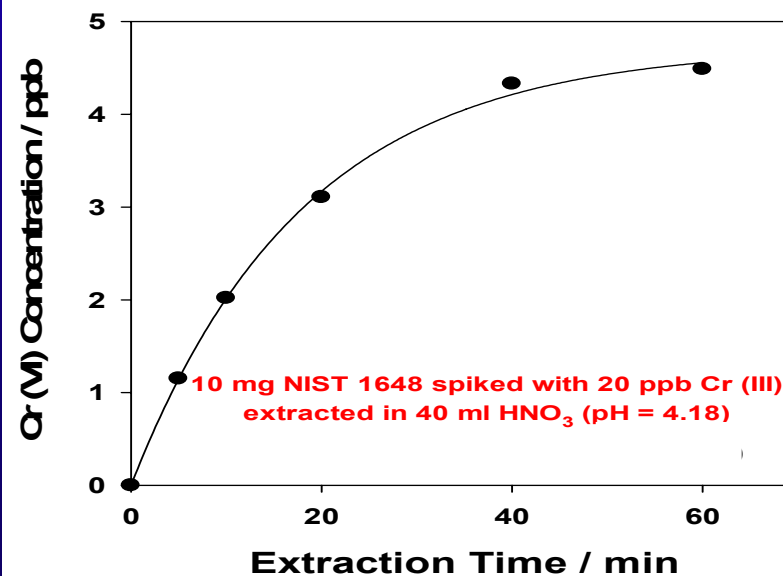
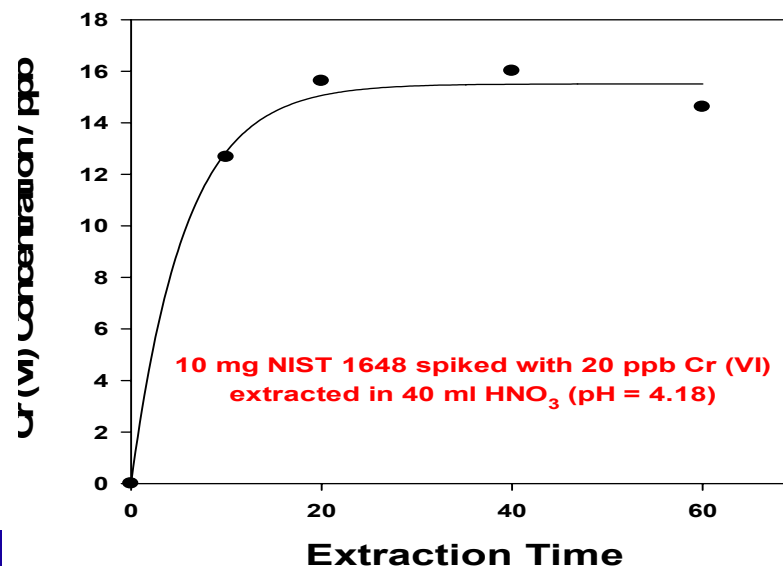
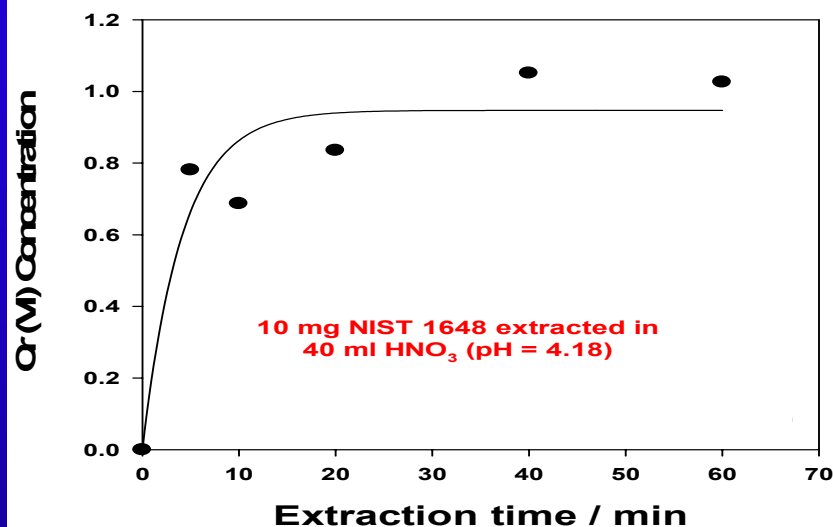
Group C: Cr (III) at pH=8.5 in  $\text{NH}_4\text{OH}$ , pH=10-14 in  $\text{NH}_4\text{OH-NaOH}$   
and then acidified by  $\text{HNO}_3$  to pH=1-2.

Group D: Cr (III) at pH=10-12 in  $\text{NH}_4\text{OH-NaOH}$ , then acidified by  $\text{HNO}_3$  to pH=2-4.

# Extraction Temperature and Time

- Extract time:
  - 5, 10, 20, 40, 60 minutes
- Extraction temperature
  - 0°C, room temperature, 60°C
- Sample: NIST 1648 particulate matter +Cr(VI)+Cr(III)+ Teflon or NaHCO<sub>3</sub> pretreated cellulose filter.

# Extraction Time and Cr(VI) Stability



➤ Optimal extraction time is 40 min.

➤ No significant conversion was observed from Cr(VI)<sup>53</sup> to Cr(III)<sup>50</sup>, but Cr(III)<sup>50</sup> is not stable during extraction.

Note: Isotopes were spiked on the a filter coated with 1648 particles and then processed.

## Field Evaluation during UCAMPP

- Precision – duplicate samples of both  $^{52}\text{Cr(VI)}$  (nature abundant) and  $^{53}\text{Cr(VI)}$  (isotope)
- Inter-conversion rate
- Recovery

## Relative Abs. Percent Difference [Cr(VI)]

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	<b>N</b>	<b>mean</b>	<b>std</b>	<b>median</b>
<b>52Cr(VI)</b>	21	16%	15%	15%
<b>53Cr(VI)</b>	23	12%	16%	7%

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**MDL: 0.18 ng/m<sup>3</sup> (0.86 ng/mL)**

## Inter-conversion and Recovery Rate [<sup>53</sup>Cr(VI) and <sup>50</sup>Cr(III)]

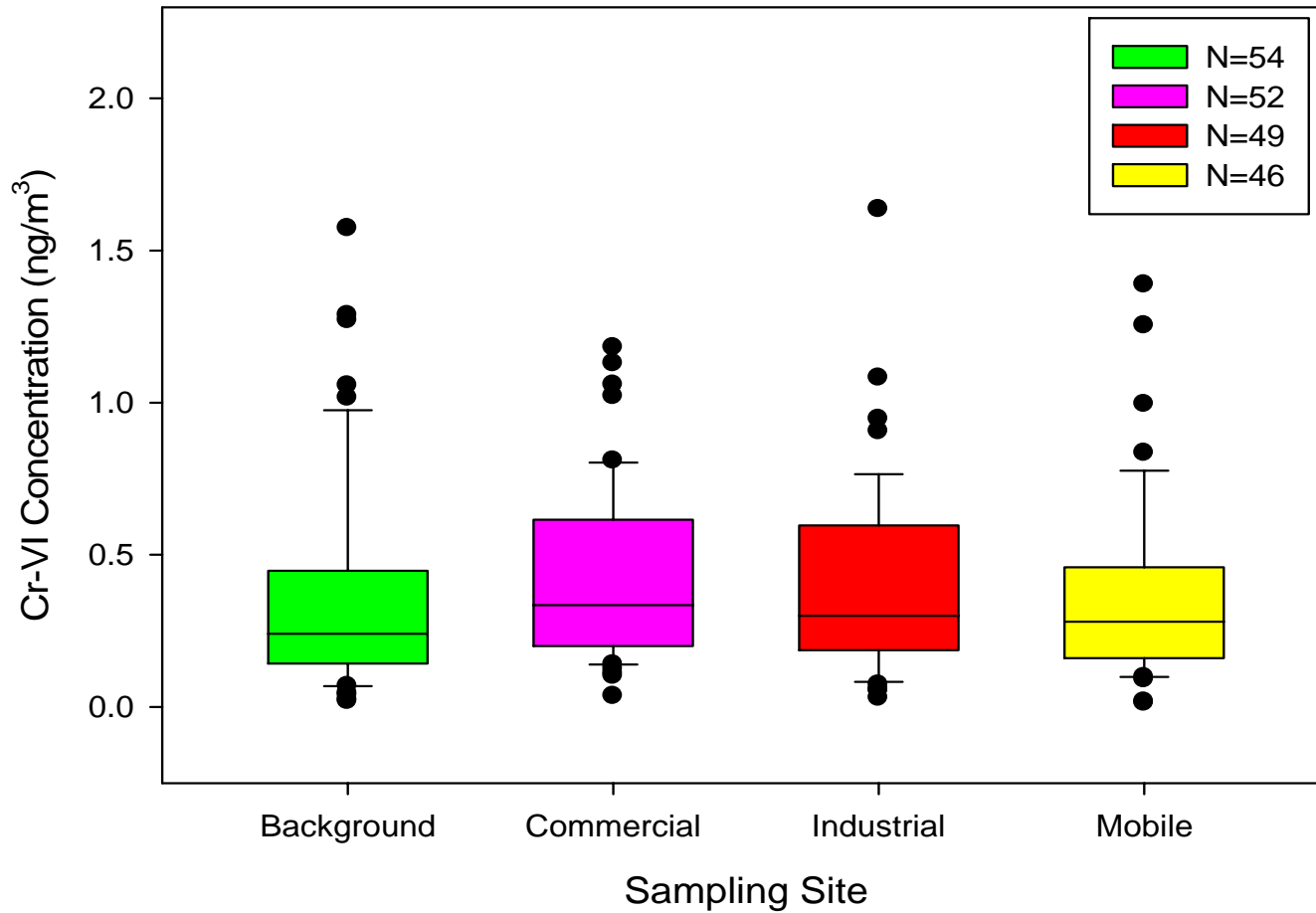
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	N	mean	std	median
<b>Interconversion Rate</b>				
Cr(VI) ->Cr(III)	77	8%	6%	8%
Cr(III)->Cr(VI)	30	18%	24%	11%
<b>Recovery</b>				
<sup>53</sup> Cr(VI)	49	73%	22%	78%
<sup>50</sup> Cr(III)	49	43%	33%	33%

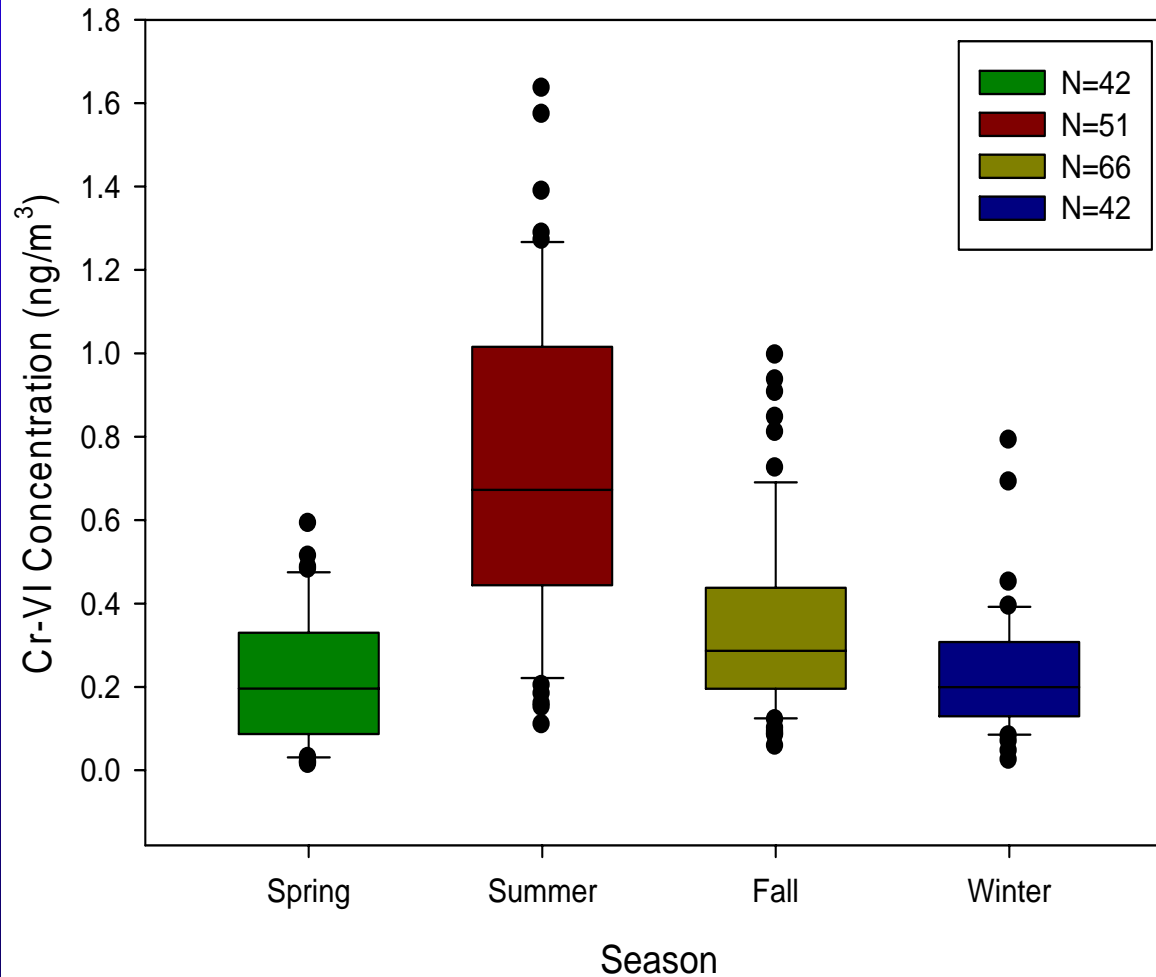
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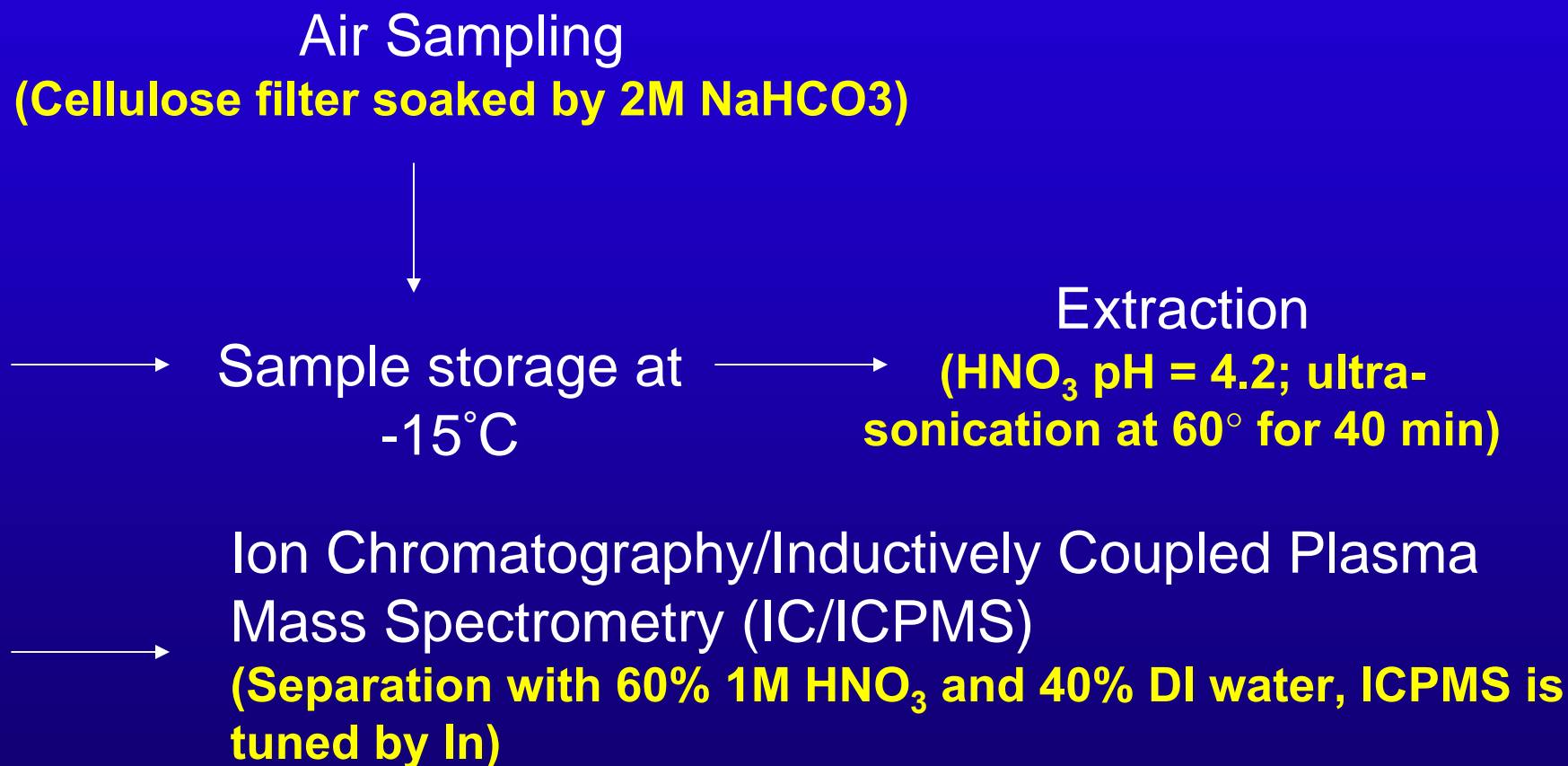
Box Plot for Cr-VI Concentrations in Air by Sampling Site



Box Plot for Cr-VI Concentrations in Air by Season



# Summary of the Cr(VI) Measurement Procedures



# Future work

- Reduce field blank Cr(VI) levels.
- Reduce/finalize inter-conversion rate between Cr(VI) and Cr(III).
- Investigate the impact of different environmental factors ( $O_3$ , UV, temperature, aerosol type, etc.) on Cr(VI) measurement.
- Inter-laboratory comparison of sampling and analytical methods.

# Acknowledgments

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